

## Ternary phase relation on preparation of YBa<sub>2</sub>Cu<sub>3</sub>O<sub>7-δ</sub> films by laser CVD

Pei Zhao<sup>a</sup>, Akihiko Ito<sup>b</sup>, Rong Tu<sup>c</sup>, Takashi Goto<sup>d</sup>

Institute for Materials Research, Tohoku University, Katahira 2-1-1, Sendai 980-8577, Japan

<sup>a</sup>zhaopei@imr.tohoku.ac.jp, <sup>b</sup>itonium@imr.tohoku.ac.jp,

<sup>c</sup>turong@imr.tohoku.ac.jp, <sup>d</sup>goto@imr.tohoku.ac.jp

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**Abstract.** *c*-axis-oriented YBa<sub>2</sub>Cu<sub>3</sub>O<sub>7-δ</sub> films were prepared by laser chemical vapor deposition (laser CVD) using Y(DPM)<sub>3</sub>, Ba(DPM)<sub>2</sub>/Ba(TMOC)<sub>2</sub> and Cu(DPM)<sub>2</sub> as precursors with enhancement by a continuous wave Nd:YAG laser. YBa<sub>2</sub>Cu<sub>3</sub>O<sub>7-δ</sub> film almost in a single phase and that with different second phases of BaCuO<sub>2</sub>, CuBaO<sub>2</sub>, BaY<sub>2</sub>O<sub>4</sub>, CuYO<sub>2</sub>, Y<sub>2</sub>O<sub>3</sub> and CuO were obtained by varying evaporation temperature of precursors. Ternary phase diagram as a function of evaporation amount of three precursors were obtained. The deposition rate of the *c*-axis-oriented YBCO film was 60 μm h<sup>-1</sup>, about 60–600 times higher than those of conventional CVD.

### Introduction

The preparation process of YBa<sub>2</sub>Cu<sub>3</sub>O<sub>7-δ</sub> (YBCO) film has attracted great attention since the discovery of high-temperature superconductor YBCO with a high  $T_c$  of about 93 K in 1987 [1–3]. YBCO film has been prepared by various processes, *i.e.*, sol-gel [4], metalorganic deposition using trifluoroacetates (TFA-MOD) [5], multilayer evaporation [6], thermal evaporation [7], pulsed laser deposition [8], sputtering [9–10], molecular beam deposition [11] and metalorganic chemical vapor deposition (MOCVD) [12]. MOCVD is a promising process for the preparation of YBCO film due to the high controllability on orientation, deposition for large area and complex shape and relatively high deposition rate ( $R_{dep}$ ) [12–20]. In MOCVD, however, YBCO film should be prepared at more higher  $R_{dep}$  and lower deposition temperature for practical applications in particular on metal tape substrate.

Several enhanced MOCVD such as magnetic-field, plasma and photo-enhanced MOCVD processes have been developed for the preparation of YBCO film. A high magnetic field enhanced MOCVD has been used to prepare YBCO film, which showed a high  $T_c$  of 88 K and a high  $J_c$  of  $1.1 \times 10^5$  A cm<sup>-2</sup>. The  $R_{dep}$  of 0.3 μm h<sup>-1</sup>, however, was not effectively improved [21–22]. Although YBCO film with high  $T_c = 90$  K and  $J_c = 3.3 \times 10^6$  A cm<sup>-2</sup> (at 77 K) has been obtained by plasma-enhanced MOCVD, the  $R_{dep}$  was scarcely improved to 1.1 μm h<sup>-1</sup> [23].

Laser is also able to enhance MOCVD process (termed laser CVD), yielding significant high deposition rate and orientation. We have prepared various oxide and non-oxide films, such as yttria stabilized zirconia (YSZ) [24], Al<sub>2</sub>O<sub>3</sub> [25] and YBCO [26] by laser CVD. In the present study, we investigate a ternary phase relation on preparation of YBCO films on Al<sub>2</sub>O<sub>3</sub> polycrystalline substrate by laser CVD.

### Experimental

YBCO films were prepared on polycrystalline Al<sub>2</sub>O<sub>3</sub> substrates (10 mm × 10 mm × 2.5 mm) by

laser CVD using  $\text{Y}(\text{DPM})_3$ ,  $\text{Ba}(\text{DPM})_2/\text{Ba}(\text{TMOD})_2$  and  $\text{Cu}(\text{DPM})_2$  (DPM; dipivaloy methanate and TMOD; 2,2,6,6-tetramethyl-1,3,5-octanedionato) as precursors. A mixture of  $\text{Ba}(\text{DPM})_2$  and  $\text{Ba}(\text{TMOD})_2$  with a molar ratio of 4 to 1 was used to suppress the decomposition of  $\text{Ba}(\text{DPM})_2$  at the temperatures over their eutectic point and to vaporize at a constant rate. A schematic of the laser CVD apparatus has been reported elsewhere [26]. A continuous wave Nd:YAG laser (wavelength: 1064 nm) with laser power ( $P_L$ ) from 50 to 200 W was employed. The laser beam was defocused up to 20 mm in diameter to irradiate the whole substrate and was introduced through a quartz window at an incident angle of  $30^\circ$  to the substrate. The  $\text{Al}_2\text{O}_3$  substrate was heated on a heating stage at pre-heating temperatures ( $T_{\text{pre}}$ ) of 673–873 K. The deposition temperature ( $T_{\text{dep}}$ ) was measured with a thermocouple inserted into a slot 1.5 mm in depth at the back side of the substrate. The flow rates of Ar and  $\text{O}_2$  gases were  $8.3 \times 10^{-7}$  and  $3.3 \times 10^{-6} \text{ m}^3 \text{ s}^{-1}$ , respectively. The composition of the YBCO film was controlled by the vaporization temperature ( $T_{\text{vap}}$ ) of the Y ( $T_Y$ ), Ba ( $T_{\text{Ba}}$ ), and Cu ( $T_{\text{Cu}}$ ) precursor at the range of  $T_Y = 450\text{--}493 \text{ K}$ ,  $T_{\text{Ba}} = 580\text{--}623 \text{ K}$  and  $T_{\text{Cu}} = 490\text{--}533 \text{ K}$ , respectively. The temperature of all the gas lines was maintained at 623 K to prevent the condensation of precursor. The total pressure was held at 1 kPa. The deposition was conducted for 600 s. The distance between the nozzle and the substrate was 28 mm. The crystal phase was identified by X-ray diffraction (XRD; Rigaku RAD-2C) and the microstructure was observed by a scanning electron microscope (SEM; Hitachi S-3100H). The composition was investigated by an electron probe micro analyzer (EPMA).

## Results and discussion

Figure 1 shows the XRD patterns of the as-deposited films. At  $T_Y = 473 \text{ K}$ ,  $T_{\text{Ba}} = 583 \text{ K}$  and  $T_{\text{Cu}} = 533 \text{ K}$ ,  $\text{CuYO}_2$  film with a little amount of  $\text{CuO}$  was obtained (Fig. 1(a)). *c*-axis-oriented YBCO film with second phases of  $\text{Y}_2\text{O}_3$ ,  $\text{CuO}$  and  $\text{BaCuO}_2$  was obtained at  $T_Y = 473 \text{ K}$ ,  $T_{\text{Ba}} = 613 \text{ K}$  and  $T_{\text{Cu}} = 516 \text{ K}$  (Fig. 1(b)) and at  $T_Y = 453 \text{ K}$ ,  $T_{\text{Ba}} = 603 \text{ K}$  and  $T_{\text{Cu}} = 513 \text{ K}$  (Fig. 1(c)). At  $T_Y = 453 \text{ K}$ ,  $T_{\text{Ba}} = 613 \text{ K}$  and  $T_{\text{Cu}} = 513 \text{ K}$ , film with a mixture phase of  $\text{BaY}_2\text{O}_4$  and  $\text{BaCuO}_2$  was obtained (Fig. 1(d)). YBCO film almost entirely in a single phase was obtained at  $T_Y = 473 \text{ K}$ ,  $T_{\text{Ba}} = 600 \text{ K}$  and  $T_{\text{Cu}}$

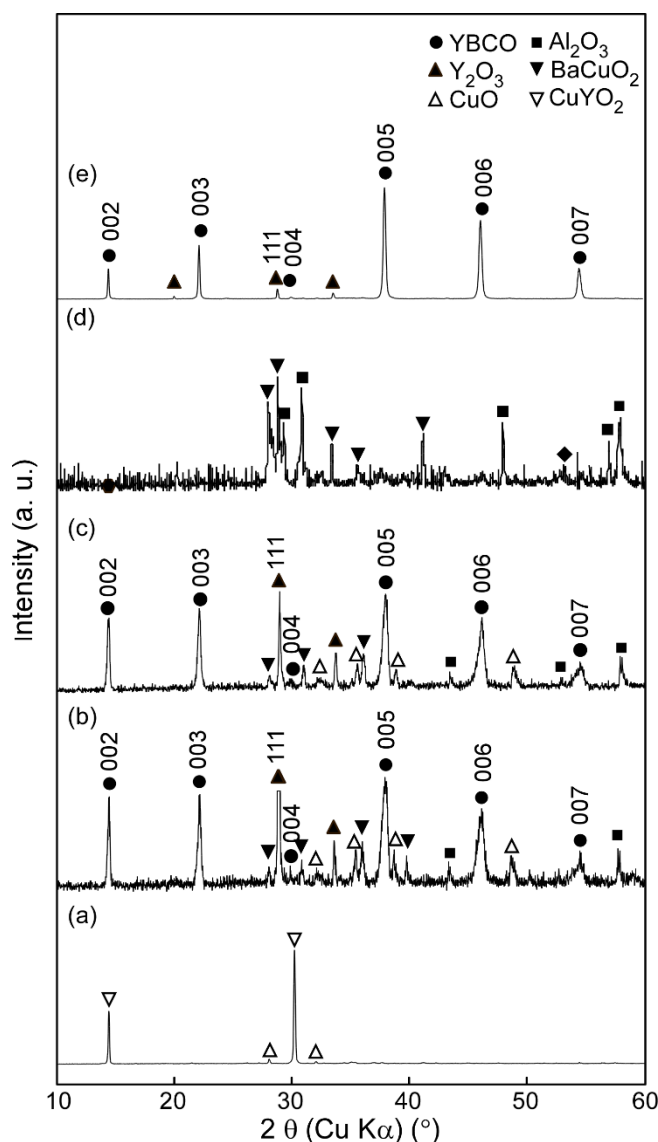


Fig. 1 XRD patterns of the YBCO film prepared at  $T_Y = 473 \text{ K}$ ,  $T_{\text{Ba}} = 583 \text{ K}$  and  $T_{\text{Cu}} = 533 \text{ K}$  (a),  $T_Y = 473 \text{ K}$ ,  $T_{\text{Ba}} = 613 \text{ K}$  and  $T_{\text{Cu}} = 516 \text{ K}$  (b),  $T_Y = 453 \text{ K}$ ,  $T_{\text{Ba}} = 603 \text{ K}$  and  $T_{\text{Cu}} = 513 \text{ K}$  (c)  $T_Y = 453 \text{ K}$ ,  $T_{\text{Ba}} = 613 \text{ K}$  and  $T_{\text{Cu}} = 513 \text{ K}$  (d), and  $T_Y = 473 \text{ K}$ ,  $T_{\text{Ba}} = 600 \text{ K}$  and  $T_{\text{Cu}} = 528 \text{ K}$  (e).

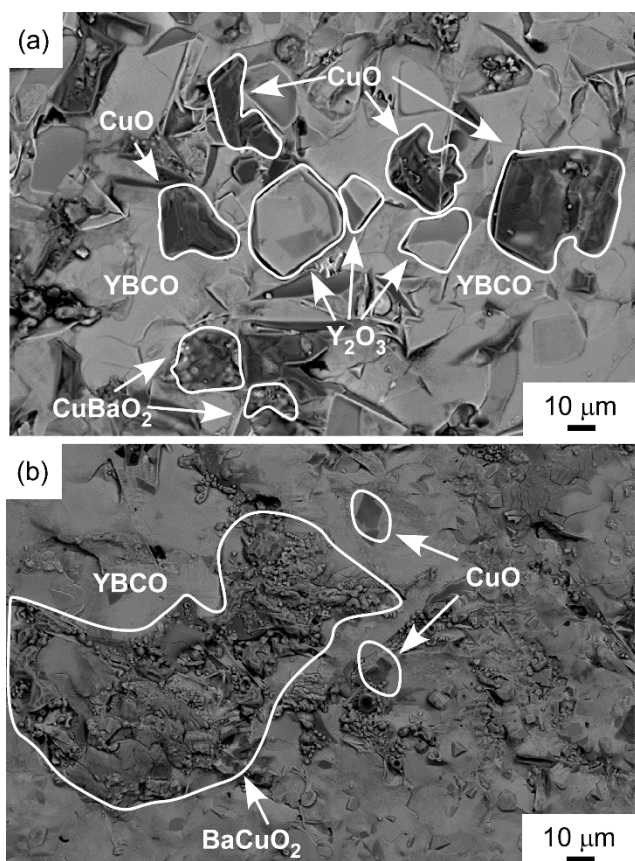


Fig.2 Surface BSE image of Y-B-C-O film prepared at  $T_Y = 473$  K,  $T_{Ba} = 613$  K and  $T_{Cu} = 516$  K (a) and  $T_Y = 453$  K,  $T_{Ba} = 603$  K and  $T_{Cu} = 513$  K (b).

= 528 K (Fig.1 (e)).

Figure 2 shows surface backscattering electron (BSE) image of YBCO film prepared at  $T_Y = 473$  K,  $T_{Ba} = 613$  K and  $T_{Cu} = 516$  K (a) and that prepared at  $T_Y = 453$  K,  $T_{Ba} = 603$  K and  $T_{Cu} = 513$  K (b). BSE image of the film showed a dense microstructure with some particles distributing in the surface (Fig. 2(a)). YBCO and  $Y_2O_3$  grains showed gray color and the  $CuBaO_2$  and  $CuO$  grains showed dark color in the BSE image (Fig. 2(a)). The  $Y_2O_3$  grains showed hexangular or pyramid shape.  $CuBaO_2$  and  $CuO$  grains showed no regular shape.  $CuBaO_2$  grains showed melted feature, which was owing to the higher deposition temperature than melting point of  $CuBaO_2$  (Fig. 2(a)). Dense microstructure was observed in the YBCO film prepared at  $T_Y = 453$  K,  $T_{Ba} = 603$  K and  $T_{Cu} = 513$  K (Fig. 2(b)).  $BaCuO_2$  phase exhibited cracked microstructure and the  $CuO$  grains showed rhombic and rectangular shapes. Both  $BaCuO_2$  and  $CuO$  phases exhibited dark color in the BSE image (Fig. 1).

Figure 3 shows surface and SEM image of *c*-axis oriented YBCO film prepared at  $T_Y = 473$  K,  $T_{Ba} = 600$  K and  $T_{Cu} = 528$  K. The surface of *c*-axis oriented YBCO film consisted of rectangular grains about 30  $\mu m$  in size, which were flat and attributed to *c*-plane of YBCO. The deposition rate of present film was  $60 \mu m h^{-1}$ , which is about 60–600 times higher than those of conventional MOCVD.

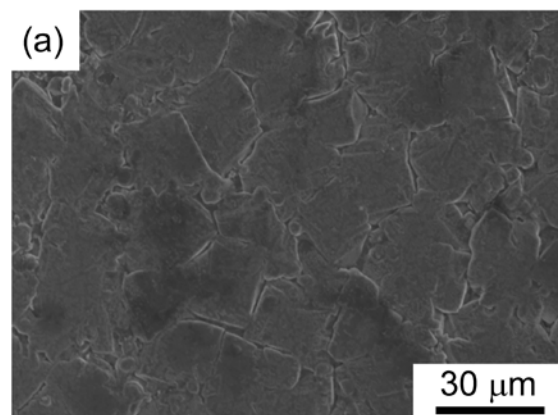


Fig. 3 Surface SEM image of *c*-axis oriented YBCO film prepared at  $T_Y = 473$  K,  $T_{Ba} = 600$  K and  $T_{Cu} = 528$  K.

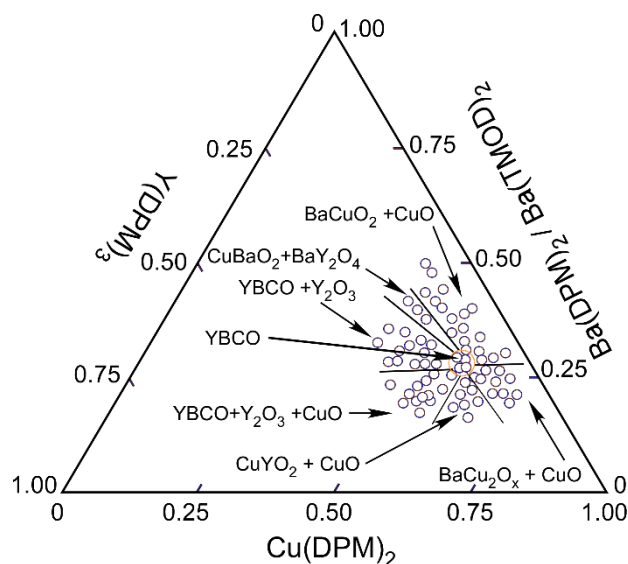


Fig. 4 Ternary phase relation as a function of evaporation amount ratio of each precursor.

Figure 4 summarizes a ternary phase relation as a function of evaporation amount ratio of Y, Ba and Cu precursors. YBCO with different phases of BaCuO<sub>2</sub>, CuBaO<sub>2</sub>, BaY<sub>2</sub>O<sub>4</sub>, CuYO<sub>2</sub>, Y<sub>2</sub>O<sub>3</sub> and CuO were obtained by changing the vaporization temperature of the Y, Ba and Cu precursors in the range of 450–493 K, 580–623 K and 490–533 K, respectively. YBCO film almost in a single phase can be prepared at  $T_Y = 473$  K,  $T_{Ba} = 600$  K and  $T_{Cu} = 528$  K.

## Summary

A ternary phase relation as a function of evaporated amount ratio of precursors was obtained by optimizing evaporation temperature of precursors. A highly *c*-axis-oriented YBCO film almost in single phase was obtained at  $T_Y = 473$  K,  $T_{Ba} = 600$  K and  $T_{Cu} = 528$  K. YBCO film had relatively large, rectangular-shaped grains about 30 μm in size. The  $R_{dep}$  of the *c*-axis-oriented YBCO film was 60 μm h<sup>-1</sup>, about 60–600 times higher than those of conventional MOCVD.

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