This is the accepted version of the article submitted to Journal of Alloys and Compounds, which has been published in final form at 10.1016/j.jallcom.2009.09.088.

Texture and Orientation Characteristics of α-Al_2O_3 Films Prepared by Laser Chemical Vapor Deposition using Nd:YAG laser

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Abstract

α-Al_2O_3 films were prepared by laser chemical vapor deposition (LCVD) and the effects of precursor vaporization temperature (T_{vap}), total chamber pressure (P_{tot}), laser power (P_L) and deposition temperature (T_{dep}) on the phase, orientation and texture of Al_2O_3 film were investigated. At P_{tot} = 0.93 kPa, α-Al_2O_3 films were obtained in the region of T_{vap} > 423 K and T_{dep} > 1100 K. The orientation of α-Al_2O_3 film changed from (110) to (012) to (104) to (006) with increasing P_{tot}. Porous α-Al_2O_3 films were formed at high T_{vap} (443 K) and low P_{tot} (0.47 kPa). At T_{vap} = 413 K, α-Al_2O_3 film had hexagonal and rectangular plate-like grains with finely faceted edges. With increasing P_{tot} = 0.93 to 1.4 kPa, (006)-oriented α-Al_2O_3 film with a hexagonal terrace texture was obtained.

Keywords: laser CVD; Alumina; Coating; Orientation; Microstructure

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A. Ito, H. Kadokura, T. Kimura, T. Goto: Texture and orientation characteristics of α-Al_2O_3 films prepared by laser chemical vapor deposition using Nd:YAG laser, *Journal of Alloys and Compounds* **489(2)** 469–474 (2010). doi:10.1016/j.jallcom.2009.09.088

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1. Introduction

Alumina (Al_2O_3) and titanium carbonitride (Ti(C, N)) are widely used as hard-coating materials. Since Al_2O_3 is characterized by high hardness and low thermal conductivity at high temperatures, the cutting efficiency and lifetime of cemented carbide (WC–Co) tools with an Al_2O_3 coating can be improved [1]. Moreover, since WC-Co tools should be able to withstand high temperatures and harsh stress conditions, when the dry cutting process with higher speed is employed, a high-performance hard coating is strongly demanded [2].

A low temperature deposition technique with phase control is required in Al_2O_3 coating due to the many polymorphs of Al_2O_3 [3]. In particular, among these polymorphs, corundum structure (α phase) appears to be the only thermodynamically stable phase with excellent hardness at high temperatures. Although α-Al_2O_3 is widely used for hard-coating material [4], the deposition temperature of α-Al_2O_3 coating by conventional thermal chemical vapor deposition (CVD) is about 1273 K. Such high temperature restricts the selection of substrate to only a few materials with a high thermal stability. Metastable phases such as the γ- and κ-phases would have to be prepared at lower deposition temperatures. However, they may transform to the α-phase above 1000 K, resulting in volume change and cracking or delamination of the Al_2O_3 coating during the cutting process [5,6].

Several characteristics of α-Al_2O_3 coating such as hardness and roughness could depend on the crystallographic orientation and texture. Ruppi *et al*. prepared (012), (104), (001)-oriented α-Al_2O_3 films on Ti(C, O) and Ti(C, N) intermediate layers deposited on WC-Co substrate and reported that the films with (006) and (104) orientations showed higher hardness and modulus than those with random or the other orientations [4,7–9]. It is commonly understood that the crystal orientation depends not only on the substrate but also on the deposition temperature and the total pressure in the CVD process. Since α-Al_2O_3 film in a single phase can only be prepared at a high deposition temperature in a limited range by conventional thermal CVD, the relationship among deposition conditions, orientation and texture has not been studied in a wide range of conditions [1,10]. In order to develop an α-Al_2O_3 coating with control of phase, orientation and texture, a low temperature α-Al_2O_3 coating could be promising as a high-performance coating for cutting tools.

We have studied low temperature and high-speed deposition of thick Y_2O_3 and TiO_2 films with control of their orientations by laser CVD (LCVD) using a high-power Nd:YAG laser (260 W) [11,12]. In this study, we prepared α-Al_2O_3 films by LCVD and investigated the effect of deposition conditions on the phase, orientation and texture of Al_2O_3 film.

2. Experimental procedure

Figure 1 shows a schematic of the LCVD apparatus and Table 1 lists the deposition conditions. Aluminium tri-acetylacetonate ($Al(acac)_3$) was used as the precursor. $Al(acac)_3$ was vaporized at vaporization temperature (T_{vap}) of 453 K and carried into the CVD chamber with Ar gas. O_2 gas and

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$Al(acac)_3$ vapor were separately introduced into the CVD chamber through a double-tube nozzle. The gas flow rates of Ar and O_2 gases were maintained at 3.3×10^{-6} m^3 s^{-1} (200 sccm). The total pressure (P_{tot}) ranged from 0.47 to 2.33 kPa. Yttria-stabilized zirconia (YSZ) plates (12.5 × 12.5 × 1.0 mm) were used as the substrate, and the YSZ substrate was heated on a hot stage at pre-heating temperatures (T_{pre}) from room temperature to 973 K. The distance between the nozzle and the substrate was fixed at 28 mm. An Nd:YAG laser in a continuous wave mode (wavelength: 1064 nm) was collimated by lenses to irradiate the whole substrate and introduced into the CVD chamber through a quartz window. The laser power (P_L) was changed up to 260 W. The deposition temperature (T_{dep}) was measured with a thermocouple inserted into the back side of the substrate.

The crystal phase was determined by X-ray diffraction (θ-2θ XRD; Rigaku RAD-2C). The surface and cross-sectional texture was observed by scanning electron microscopy (SEM; Hitachi S-3100H). The deposition rate (R_{dep}) was calculated from the thickness and the deposition time.

3. Results and discussion

In common CVD process, T_{dep} is obviously one of the most dominant parameters and usually determined by heating with an electric current or by a high-frequency electromagnetic wave, etc. In the LCVD process, on the other hand, the T_{dep} can be determined for various parameters such as P_L, T_{pre}, P_{tot} and T_{vap}. For example, the P_L dependence on T_{dep} at various T_{vap} for T_{pre} = 873 K and P_{tot} = 0.97 kPa is depicted in Fig. 2. By laser irradiation, the T_{dep} increased by 150 to 200 K from T_{pre}, and the temperature increase from T_{pre} to T_{dep} became more than 400 K when the P_L was increased to 260 W. The temperature increase tended to be greater at higher T_{vap}. This might be caused by the increase in exothermic heat by the reaction between the precursor and O_2 gases as well as by the increase in significant plasma formation by laser irradiation of the gases.

Figure 3 shows the XRD patterns of Al_2O_3 films prepared at P_L = 192 W. At P_{tot} = 0.93 kPa and T_{vap} = 423 K (T_{dep} = 1210 K), α-Al_2O_3 film in a single phase with slight (012) and (104) orientation was obtained (Fig. 3(a)). At T_{vap} = 443 K, a peak around 2θ = 41.7° (d = 0.217 nm) from the (006) plane was identified (Fig. 3(b)), its intensity increasing at P_{tot} = 1.4 kPa (Fig. 3(c)). The diffraction intensity of the (006) plane is only 2.0 compared with 100 for the (104) plane according to the JCPDS card (No. 46-1212). Therefore, the intensity of the (006) peak cannot be identified in common α-Al_2O_3. The Harris texture coefficient (TC) can be used to quantitatively estimate the orientation, particularly for a small diffraction peak [13].

$$TC(hkl) = N \frac{I_m(hkl)/I_0(hkl)}{\sum I_m(hkl)/I_0(hkl)}, \quad (1)$$

where I_m(hkl) and I_0(hkl) are the intensity from the (hkl) plane measured in the present study and that reported in JCPDS card (No. 46-1212), respectively. (012), (104), (110), (006), (113), (024), (116), (214) and (300) planes were used for the calculation (N = 9). $TC(hkl)$ has a value between 0 to N depending

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on the orientation degree of the (*hkl*) plane. The *TC* of a non-oriented plane should be 1.0. If the TC is more than 1.0, we can define it as being oriented. If the TCs of more than two planes increase simultaneously, we can define them as being co-oriented. Figure 4 shows the *TC* for (110), (012), (104) and (006) planes of α-Al_2O_3 films prepared at P_L = 192 W and various P_{tot} and T_{vap}. The α-Al_2O_3 film prepared at P_{tot} = 0.47 kPa and T_{vap} = 423 K was (110) and (012) co-oriented (open circles in Fig. 4), and that prepared at P_{tot} = 0.93 kPa was (012) and (104) co-oriented (filled circles in Fig. 4). At P_{tot} = 0.93 kPa and T_{vap} = 443 K, the α-Al_2O_3 film changed to (104) and (006) co-oriented (open triangles in Fig. 4). The α-Al_2O_3 film prepared at T_{vap} = 443 K and P_{tot} = 1.4 kPa was significantly (006) oriented (filled triangles in Fig. 4).

Figure 5 depicts the effect of T_{dep} and T_{vap} on the phase formation of Al_2O_3 film. At T_{dep} < 1060 K, γ-Al_2O_3 films in a single phase were formed independently of T_{vap}. Al_2O_3 films with a γ + θ mixture phase were formed at T_{vap} = 403 K and T_{dep} > 1070 K, whereas Al_2O_3 films with an α + θ mixture phase were formed at T_{vap} = 413 K and T_{dep} > 1116 K. At T_{vap} > 423 K, the phase changed from a γ to a γ + α mixture to an α phase with increasing T_{dep}.

Figure 6 depicts the effect of P_{tot} and T_{vap} on the phase and orientation of Al_2O_3 film. At P_{tot} = 0.47 kPa, α-Al_2O_3 film was prepared at T_{vap} > 413 K. With increasing T_{vap}, the orientation changed from (110) to (110) and (012) co-orientation to (104) and (012) co-orientation. At P_{tot} = 0.93 kPa, α-Al_2O_3 film was also prepared at T_{vap} > 423 K, and (012) and (104) co-oriented and (104) and (006) co-oriented α-Al_2O_3 films were prepared. A significant (006)-oriented α-Al_2O_3 film was obtained at P_{tot} = 1.4 kPa and T_{vap} = 443 K. At P_{tot} > 1.9 kPa and T_{vap} < 423 K, γ-Al_2O_3 film in a single phase was formed, and an α phase appeared with a γ phase at T_{vap} > 433 K. It can be summarized that at T_{vap} > 423 K, the Al_2O_3 film prepared at P_{tot} < 0.93 kPa tended to be α phase, while the Al_2O_3 film at P_{tot} > 1.4 kPa tended to be γ phase. At lower T_{vap}, a θ phase appeared with the α and γ phases. The orientation of α-Al_2O_3 film changed from (110) to (012) to (104) to (006) with increasing P_{tot}.

Figure 7 shows the surface SEM images of Al_2O_3 films prepared at various T_{vap}, P_{tot} and P_L. (104) and (006) co-oriented α-Al_2O_3 film prepared at T_{vap} = 443 K, P_{tot} = 0.93 kPa and P_L = 192 W had hexagonal faceted grains slightly inclined to the substrate (Fig. 7(a)). (006)-oriented Al_2O_3 film was obtained at P_{tot} = 1.4 kPa, and a hexagonal terrace texture developed on the surface of the grains (Fig. 7(b)). When the P_L increased to 238 W at P_{tot} = 0.93 kPa, rounded hexagonal and rectangular-shaped grains were observed (Fig. 7(c)). At T_{vap} = 443 K and P_{tot} = 0.47 kPa, porous α-Al_2O_3 film was formed (Fig. 7(d)). As in the conventional CVD process, a porous structure was often obtained under a high supersaturation condition of source gases. In the present study, the gas supply rate at T_{vap} = 443 K and P_{tot} = 0.47 kPa increased to a level 2 times higher than those of the other deposition conditions. When the T_{vap} decreased to 433 K, hexagonal and rectangular plate-like grains were grown with finely faceted edges (Fig. 7(e)). At P_{tot} = 2.3 kPa, γ-Al_2O_3 film having a cauliflower-like structure in a single phase was obtained (Fig. 7(f)). A typical cross-sectional SEM image of oriented α-Al_2O_3 films is shown in Fig. 8. The α-Al_2O_3 films

with (104) and (006) co-orientation prepared at T_{vap} = 433 K, P_{tot} = 0.93 kPa and P_L = 192 W had a columnar structure. The step structure was observed at the surface of columnar grains and the feather-like texture was formed in the cross section. The thickness of the most α-Al_2O_3 films was ranging from 15 to 25 μm and the highest R_{dep} was 540 μm h^{-1} at T_{dep} = 1100 K for γ-Al_2O_3 film, and was 324 μm h^{-1} at T_{dep} = 1200 K for α-Al_2O_3 film.

Figure 9 summarizes the effect of P_{tot} and T_{vap} on the texture of Al_2O_3 films prepared at various deposition conditions. At low P_{tot} (0.47 kPa) and high T_{vap} (443 K), porous α-Al_2O_3 film was formed. With increasing P_{tot} to 1.4 kPa, the α-Al_2O_3 film had columnar grains with a hexagonal terrace texture, while with decreasing T_{vap} to 413 K, α-Al_2O_3 film had hexagonal and rectangular plate-like grains with finely faceted edges. At higher P_{tot} or lower T_{vap}, γ- and θ-phases appeared with the α-phase, and non-oriented grains were grown with a hexagonal and rectangular faceted texture. At P_{tot} > 1.4 kPa, γ-Al_2O_3 film with a cauliflower-like texture was obtained.

The orientation and texture of α-Al_2O_3 film also depended on the deposition conditions such as T_{vap} and P_{tot}. Schematics of the cross-sectional crystal structure of the (006), (104), (012) and (110) planes of α-Al_2O_3 are presented in Fig. 10. In the corundum structure of α-Al_2O_3, the Al_2O_9 octahedral dimers are joined alternately by edge- and corner-sharing in the *c*-axis direction. (012) and (104) planes intersect with the (006) plane at angles of 57.6° (Fig. 10(b)) and 38.2° (Fig. 10(c)), respectively. The orientation of α-Al_2O_3 film changed from (110) to (012) to (104) to (006) with increasing P_{tot}, as summarized in Fig. 5. It may be assumed that the longer direction of Al_2O_9 octahedral dimers (corresponding to the packing density of Al cations) decreased with increasing P_{tot} and T_{vap}. It is generally understood that the (001) orientation of hexagonal lattice would appear at a low supersaturation condition in vapor and liquid phase deposition in various systems. In the present LCVD, (006) oriented α-Al_2O_3 films were obtained under rather high P_{tot} conditions. Since homogeneous reactions in a gas phase (mainly powder formation) occurs more significantly at higher P_{tot}, resulting in depletion of source gases, supersaturation would be decreased with increasing P_{tot}. This might have caused the (006) orientation at high P_{tot}.

4. Conclusions

α-Al_2O_3 films were prepared by LCVD and the effects of deposition conditions on the phase, orientation and texture of Al_2O_3 film were investigated. At P_{tot} = 0.93 kPa, α-Al_2O_3 films were obtained in the region of T_{vap} > 423 K and T_{dep} > 1100 K. The orientation of α-Al_2O_3 film changed from (110) to (012) to (104) to (006) with increasing P_{tot}. Porous α-Al_2O_3 film was formed at high T_{vap} (443 K) and low P_{tot} (0.47 kPa). When T_{vap} decreased to 413 K, α-Al_2O_3 film had hexagonal and rectangular plate-like grains with finely faceted edges, while with increasing P_{tot} = 0.93 to 1.4 kPa, (006)-oriented α-Al_2O_3 film with a hexagonal terrace texture was obtained.

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Acknowledgement

This research was supported in part by the Global COE Program of Materials Integration, Tohoku University, and by the Asian CORE Program, Japan Society for the Promotion of Science (JSPS). This research was also performed as part of the Rare Metal Substitute Materials Development Project of the New Energy and Industrial Technology Development Organization (NEDO).

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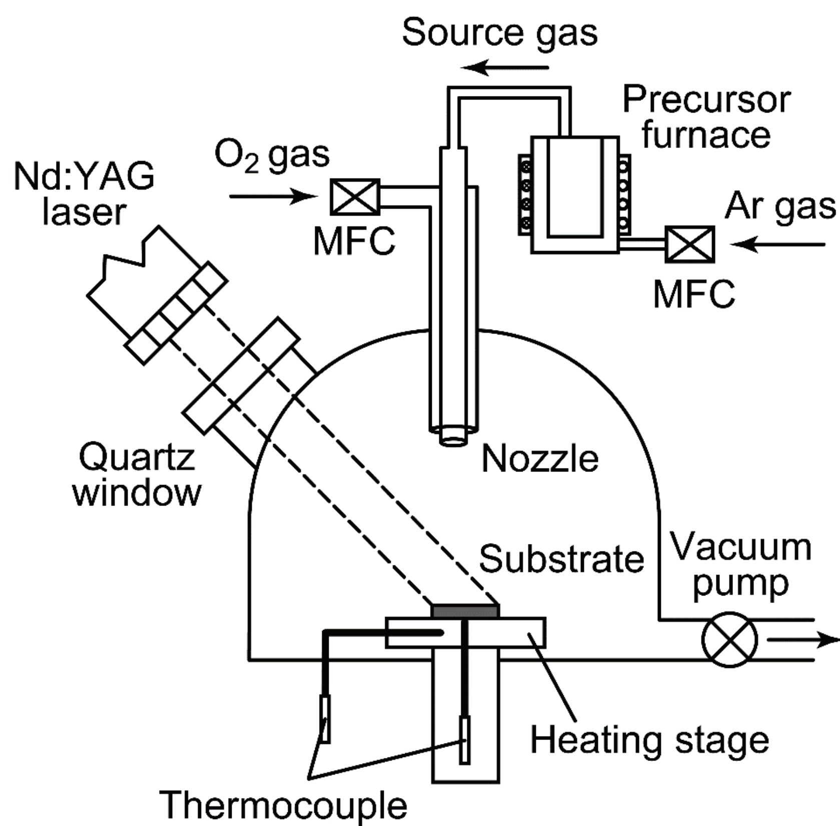

Fig. 1 Schematic of laser CVD apparatus.

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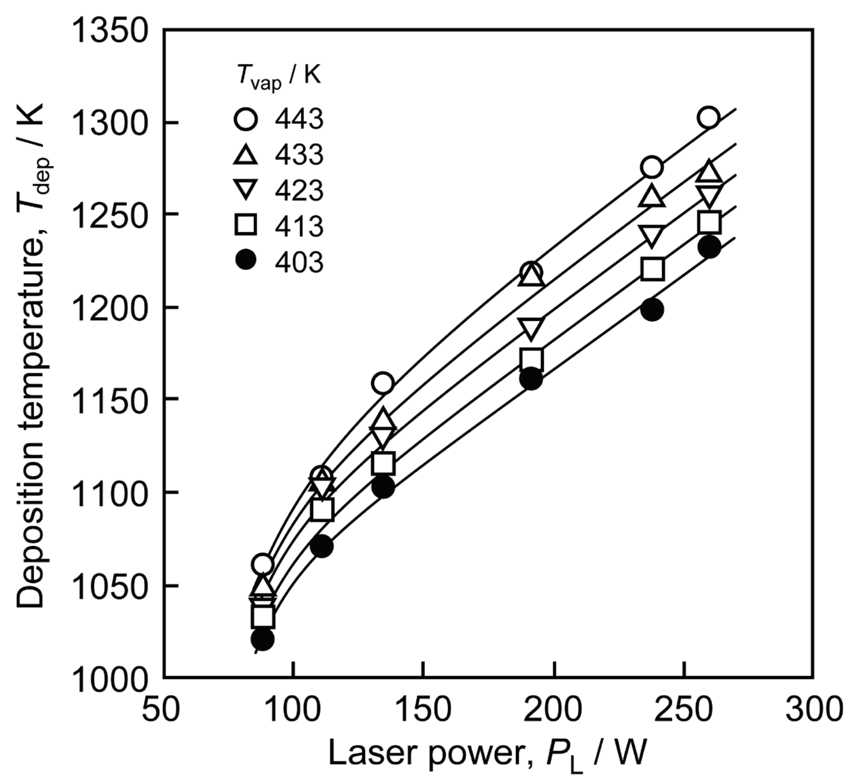

Fig. 2 P_L dependence on the T_{dep} at T_{pre} = 873 K, P_{tot} = 0.97 kPa and various T_{vap}.

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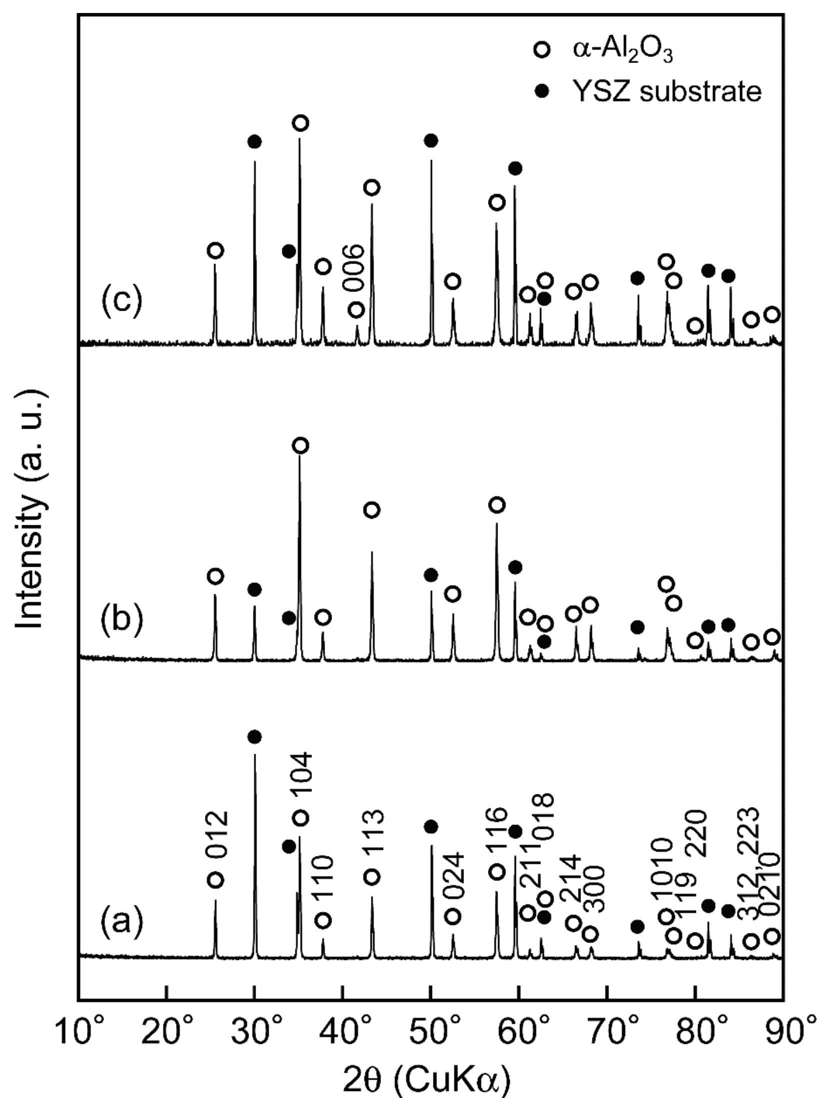

Fig. 3 XRD patterns of Al_2O_3 films prepared at P_L = 192 W, P_{tot} = 0.93 kPa and T_{vap} = 423 K (T_{dep} = 1210 K) (a), P_{tot} = 0.93 kPa and T_{vap} = 443 K (T_{dep} = 1230 K) (b), and P_{tot} = 1.4 kPa and T_{vap} = 443 K (T_{dep} = 1210 K) (c).

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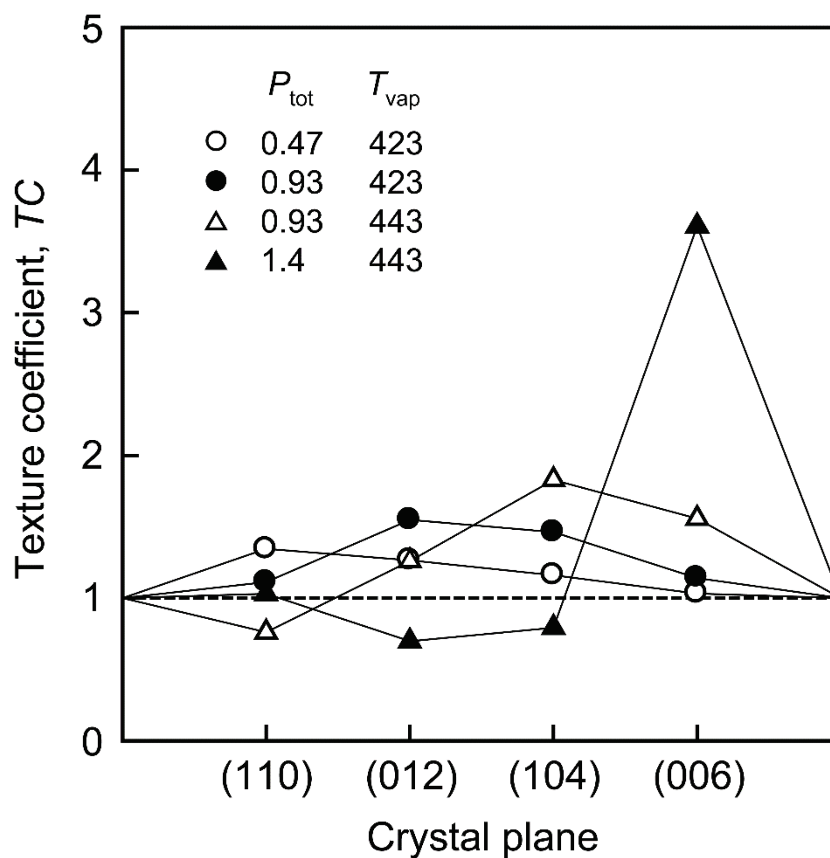

Fig. 4 Preferred orientations of Al_2O_3 films prepared at P_L = 192 W and various P_{tot} and T_{vap}: 0.47 kPa and 423 K (open circles), 0.93 kPa and 423 K (filled circles), 0.93 kPa and 443 K (open triangles), and 1.4 and 443 K (filled triangles).

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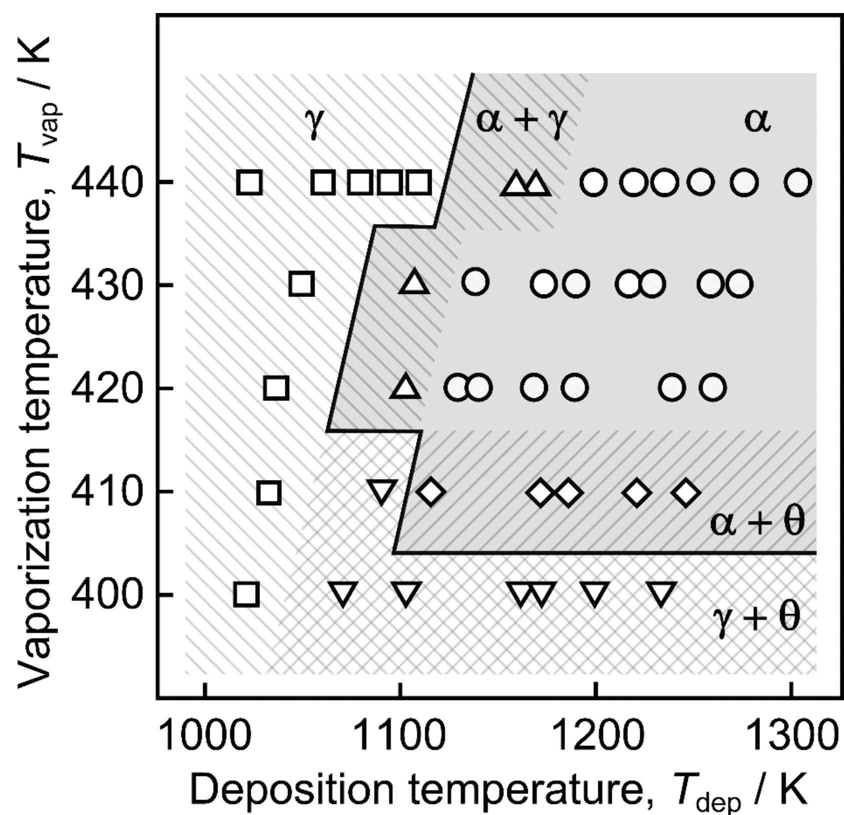

Fig. 5 Effect of T_{dep} and T_{vap} on the phase of Al_2O_3 films prepared at P_{tot} = 0.93 kPa.

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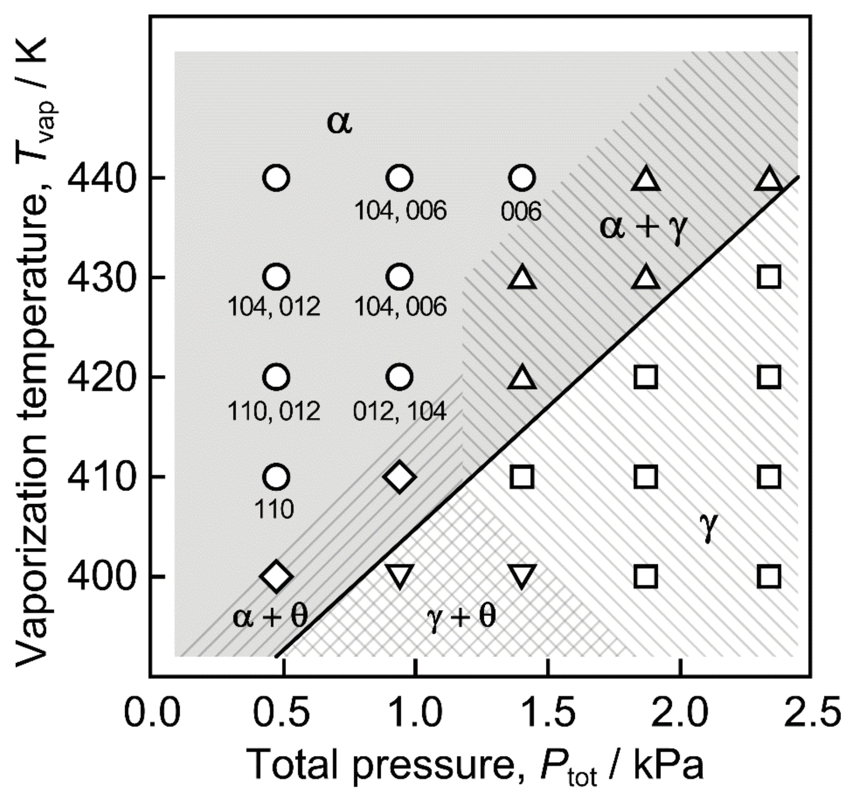

Fig. 6 Effect of P_{tot} and T_{vap} on the phase and orientation of Al_2O_3 films prepared at P_L = 192 W.

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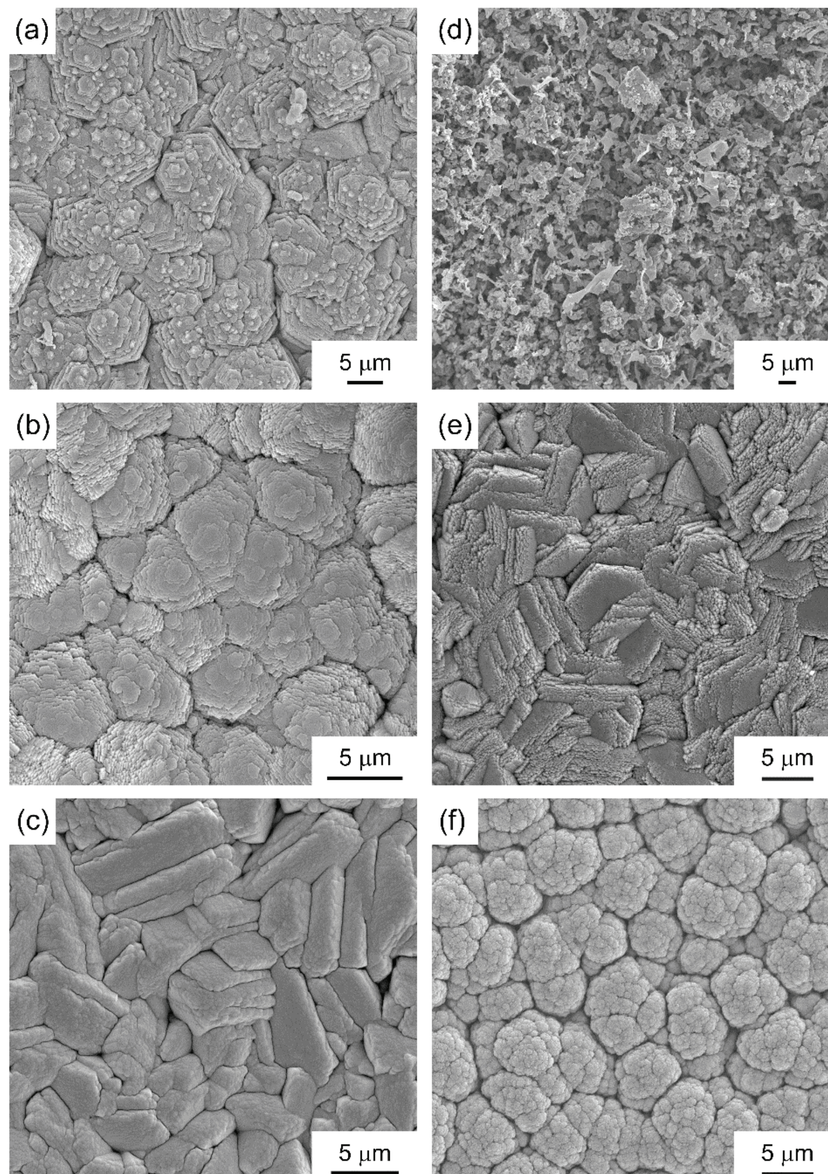

Fig. 7 Surface SEM images of Al_2O_3 films prepared at T_{vap} = 443 K and various P_{tot} and P_L: P_{tot} = 0.93 kPa and P_L = 192 W (a), P_{tot} = 1.4 kPa and P_L = 192 W (b), P_{tot} = 0.93 kPa K and P_L = 238 W (c), and at P_L = 192 W and various P_{tot} and T_{vap}: P_{tot} = 0.47 kPa and T_{vap} = 443 K (d), P_{tot} = 0.47 kPa and T_{vap} = 433 K (e), and P_{tot} = 2.3 kPa and T_{vap} = 433 K (f).

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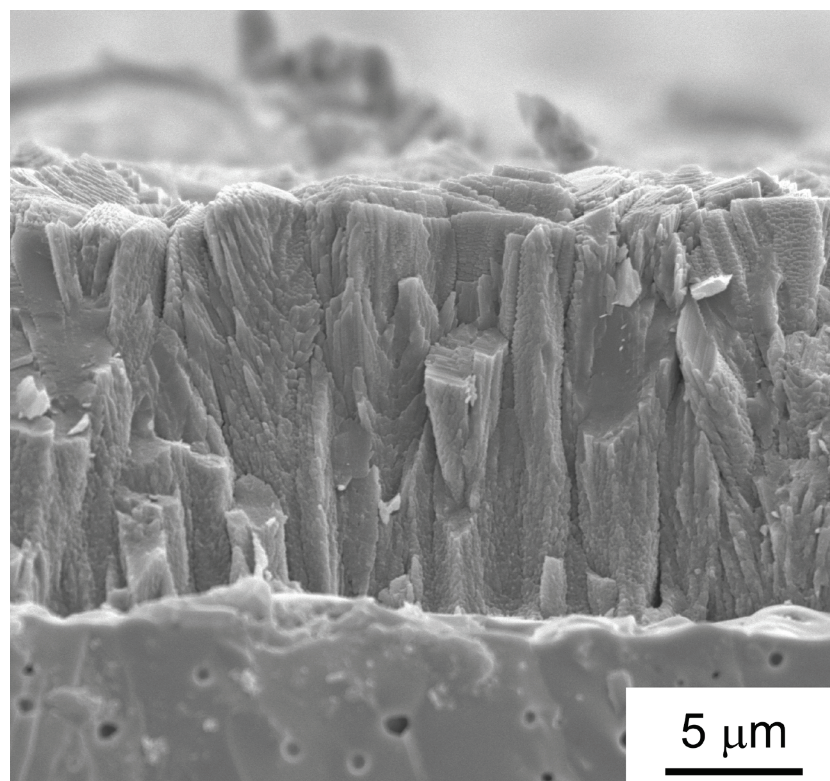

Fig. 8 Cross-sectional SEM image of α-Al_2O_3 films with (104) and (006) co-orientation prepared at T_{vap} = 433 K, P_{tot} = 0.93 kPa and P_L = 192 W.

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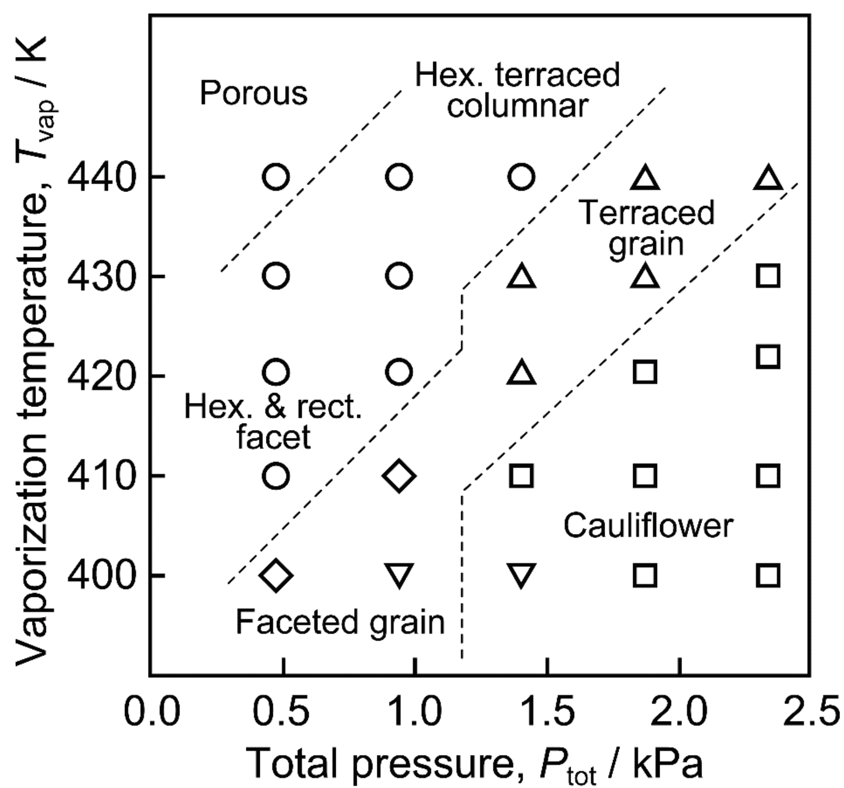

Fig. 9 Effects of P_{tot} and T_{vap} on the texture of Al_2O_3 films prepared at P_L = 192 W. Plot shapes correspond to the phase (Fig. 5).

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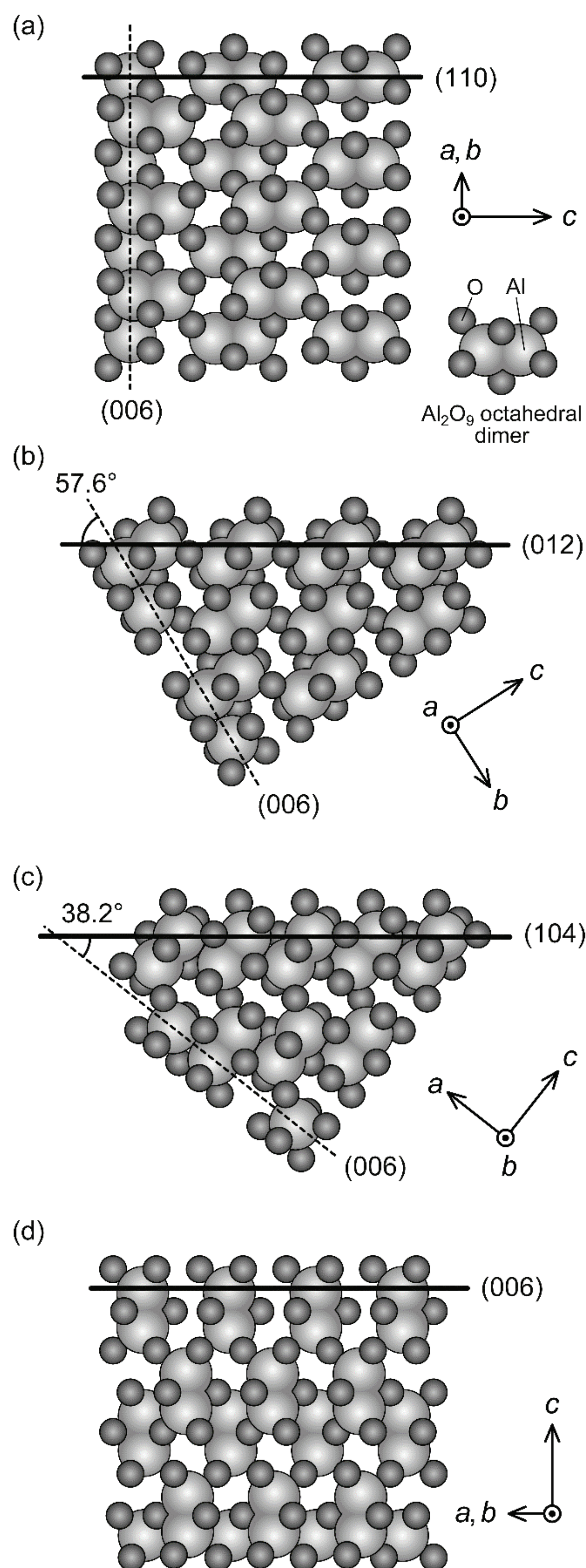

Fig. 10 Schematic of cross-sectional crystal structure of (006) (a), (104) (b), (012) (c) and (110) (d) planes of α- Al_2O_3.

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Table 1 Deposition conditions

$Al(acac)_3$ vaporization temperature (T_{vap}):	403–443 K
Substrate pre-heating temperature (T_{pre}):	873 K
Gas flow rate for both Ar and O_2 gases:	3.3×10^{-6} m^3 s^{-1}
Total pressure (P_{tot}):	0.47–2.33 kPa
Distance between nozzle and substrate:	28 mm
Laser output power (P_L):	88–236 W